JEE Main April 2025: Chemistry Preparation Checklist

	Create one-page Physical Chemistry formula sheet Prepare Organic Chemistry reaction mechanism flashcards Make Periodic Table trends quick-reference guide Set up Named Reactions summary sheet Prepare an error log notebook Download/organize past JEE Main papers (last 5 years) Establish daily study schedule with time blocks for all subjects Set up progress tracking system
Daily	y Must-Do Checklist
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	20-30 minutes formula/concept revision Complete topic of the day (60-90 minutes) Solve 15-20 practice MCQs Review mistakes from previous days (10 minutes) Update quick-reference materials with new insights Cross-check topic coverage with JEE syllabus Complete 2-minute recall after each chemistry session
Phys	sical Chemistry Priority List
	Chemical & Ionic Equilibrium
	 □ Kc, Kp relationships □ pH calculations □ Buffer solutions □ Solubility product □ Common ion effect Thermodynamics
	☐ First & Second Laws☐ Enthalpy, Entropy, Free Energy☐ Spontaneity of reactions☐ Thermochemical equations

	Electrochemistry
	☐ Electrochemical cells
	□ Nernst equation
	☐ EMF calculations
	☐ Electrode potentials
	Solutions & Colligative Properties
	☐ Raoult's law and deviations
	☐ Elevation in boiling point
	□ Depression in freezing point
	☐ Osmotic pressure
	□ van't Hoff factor
	Chemical Kinetics
	☐ Rate laws and order
	☐ Half-life calculations
	☐ Arrhenius equation
	☐ Activation energy
Orga	anic Chemistry Priority List
	General Organic Chemistry
	☐ IUPAC nomenclature
	☐ IUPAC nomenclature ☐ Isomerism (all types)
	☐ Isomerism (all types)
	☐ Isomerism (all types) ☐ Electronic effects ☐ Stability of intermediates ☐ Acidity/basicity concepts
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	☐ Isomerism (all types) ☐ Electronic effects ☐ Stability of intermediates ☐ Acidity/basicity concepts Reaction Mechanisms ☐ SN1 vs SN2
	□ Isomerism (all types) □ Electronic effects □ Stability of intermediates □ Acidity/basicity concepts Reaction Mechanisms □ SN1 vs SN2 □ E1 vs E2
	□ Isomerism (all types) □ Electronic effects □ Stability of intermediates □ Acidity/basicity concepts Reaction Mechanisms □ SN1 vs SN2 □ E1 vs E2 □ Addition reactions
	□ Isomerism (all types) □ Electronic effects □ Stability of intermediates □ Acidity/basicity concepts Reaction Mechanisms □ SN1 vs SN2 □ E1 vs E2 □ Addition reactions □ Electrophilic aromatic substitution
	☐ Isomerism (all types) ☐ Electronic effects ☐ Stability of intermediates ☐ Acidity/basicity concepts Reaction Mechanisms ☐ SN1 vs SN2 ☐ E1 vs E2 ☐ Addition reactions ☐ Electrophilic aromatic substitution Carbonyl Compounds
	□ Isomerism (all types) □ Electronic effects □ Stability of intermediates □ Acidity/basicity concepts Reaction Mechanisms □ SN1 vs SN2 □ E1 vs E2 □ Addition reactions □ Electrophilic aromatic substitution Carbonyl Compounds □ Aldehydes and Ketones
	□ Isomerism (all types) □ Electronic effects □ Stability of intermediates □ Acidity/basicity concepts Reaction Mechanisms □ SN1 vs SN2 □ E1 vs E2 □ Addition reactions □ Electrophilic aromatic substitution Carbonyl Compounds □ Aldehydes and Ketones □ Carboxylic acids and derivatives

	□ Diazonium salts
	□ Preparation and reactions
	Named Reactions
	☐ Aldol condensation
	☐ Cannizzaro reaction
	☐ Sandmeyer reaction
	☐ Wurtz reaction
	☐ Hoffmann bromamide
	☐ Reimer-Tiemann reaction
Inor	ganic Chemistry Priority List
,	, ,
	p-Block Elements
	☐ Group 15-17 elements
	Important compounds
	Oxoacids
	☐ Anomalous behavior of first elements
	Coordination Compounds
	"EAC 1
	☐ IUPAC nomenclature
	☐ Isomerism
	☐ Crystal Field Theory
	☐ Magnetic properties and color
	d & f Block Elements
	☐ Electronic configurations
	☐ Properties
	☐ Important compounds
	Periodic Properties
	Atomio/ionio radii
	☐ Atomic/ionic radii
	☐ Ionization energy
	☐ Electronegativity☐ Metallic character
	_
	Chemical Bonding
	☐ Ionic, covalent, metallic bonds
	☐ VSEPR theory
	☐ Hybridization
	☐ MOT concepts

Week	dy Progress Check
	Week 1: Completed Physical Chemistry priority topics Week 1: Completed beginning of Organic Chemistry Week 1: Taken at least one sectional test Week 2: Completed Organic Chemistry priority topics Week 2: Completed Inorganic Chemistry priority topics Week 2: Taken at least one full mock test
Mock	Test Schedule
]]]	Day 5: Physical Chemistry sectional test Day 8: Organic Chemistry sectional test Day 11: Inorganic Chemistry sectional test Day 13: Full Chemistry mock test Day 14: Error analysis and final revision
Final	3 Days Preparation
	Review all formula sheets and quick-reference materials Practice 30 mixed MCQs from all areas Revise common error patterns from your error log Review high-weightage topics one final time Prepare exam day kit (admit card, stationery, etc.) Get proper sleep and stay hydrated Brief review of most critical formulas/reactions
Exam	Day Strategy
1	Wake up early and review key formulas (30 mins) Follow first-scan approach for Chemistry section Allocate proper time: Physical (12-15 mins), Organic (10-12 mins), Inorganic (8-10 nins) Use elimination method for uncertain questions Double-check calculations before marking answers Skip and return to time-consuming questions
Mate	rials Needed
☐ F	NCERT Chemistry Textbooks (XI & XII) Formula handbook/quick reference Previous years' JEE Main question papers

	Topic-specific MCQ collections
	Index cards for flashcards
	Highlighters and sticky notes
\Box	Progress tracking notebook/app

